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Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.003 Å R factor = 0.036 wR factor = 0.100 Data-to-parameter ratio = 9.7

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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Dispiro[cyclopropane-1,1'(2'H)-cyclobuta[b]naphthalene-2',1"-cyclopropane]-3',8'-dione

The title compound, $C_{16}H_{12}O_2$, contains two independent molecules in the asymmetric unit. In both molecules, the benzoquinone system is essentially planar and the cyclobutene ring is coplanar with it; the cyclopropyl planes are orthogonal to the cyclobutene ring. The crystal packing is stabilized by $C-H\cdots O$ interactions as well as van der Waals forces.

Comment

We have recently investigated the photo-induced reactions of a series of quinones with bicyclopropylidene (Wang *et al.*, 2003; Usman *et al.*, 2002). During this investigation, we carried out photoreactions of 2-chloronaphthoquinone with bicyclopropylidene and obtained the title compound, (I), as one of the products. This compound was formed by the dehydrochlorination of the primary [2 + 2]-cycloadduct of these two reactants. An X-ray crystallographic analysis of (I) was undertaken to establish the structure and stereochemistry.



The structure of (I) consists of two crystallographically independent molecules A and B in the asymmetric unit of the non-centrosymmetric space group $P2_1$ (Fig. 1). The bond lengths and angles of A and B (Table 1) agree with each other and are within normal ranges (Allen *et al.*, 1987). In both molecules, the quinone ring is essentially planar, and the dihedral angles between this plane and that of the fused benzene ring are 3.95 (10) and 2.48 (12)° in molecules A and B, respectively. The cyclobutene plane makes dihedral angles of 3.38 (13) and 1.35 (13)° with the benzoquinone ring in A and B, in contrast with the corresponding value of 73.5 (2)° in a related structure (Wang *et al.*, 2003). This is because the C8=C11 double bond makes the cyclobutene ring and the benzoquinone system essentially coplanar.

In both A and B, the configurations of two cyclopropyl rings with respect to the cyclobutene are determined by the C16A– C9A–C10A–C14A and C15A–C9A–C10A–C13A torsion angles [-3.6 (3) and -4.5 (4)°] in A, and the corresponding C16B–C9B–C10B–C14B and C15B–C9B–C10B–C13B torsion angles [3.9 (3) and 3.2 (3)°] in B. This implies that atoms C9 and C10 are eclipsed, as observed in the related structure (Wang *et al.*, 2003). The two cyclopropyl rings are Received 7 October 2003 Accepted 13 October 2003 Online 23 October 2003



Figure 1

The structure of (I), showing 50% probability displacement ellipsoids and the atom-numbering scheme.

orthogonal to the cyclobutene ring, with dihedral angles of 89.1 (2) and 89.5 (2)° in A, 88.7 (2) and 89.7 (2)° in B. Meanwhile the two cyclopropyl rings are almost perpendicular to one another, with dihedral angles of 87.5 (3)° in A and 86.3 (3)° in B.

In the crystal structure of (I), molecules A and B are interconnected by $C15B-H15D\cdots O2A$ and $C4A-H4A\cdots O1B(1 + x, 1 + y, z)$ interactions (Fig. 2 and Table 2). The packing is stabilized by these interactions, as well as van der Waals forces.

Experimental

The title compound was prepared by photolysis of a benzene solution of 2-chloronaphthoquinone in the presence of an excess amount of bicyclopropylidene, followed by silica-gel column chromatographic separation of the reaction mixture with petroleum ether (b.p. 333–363 K)–ethyl acetate as eluants. Single crystals suitable for X-ray crystallographic analysis were grown by slow evaporation of a solution in petroleum ether (b.p. 333–363 K)–acetone (5:1, v/v).

Crystal data

$C_{16}H_{12}O_2$	$D_x = 1.309 \text{ Mg m}^{-3}$
$M_r = 236.26$	Mo $K\alpha$ radiation
Monoclinic, P2 ₁	Cell parameters from 4723
a = 8.7281 (5) Å	reflections
b = 8.2720 (4) Å	$\theta = 2.4 - 28.2^{\circ}$
$c = 16.9772 \ (9) \text{ Å}$	$\mu = 0.09 \text{ mm}^{-1}$
$\beta = 102.0473 \ (9)^{\circ}$	T = 293 (2) K
$V = 1198.74 (11) \text{ Å}^3$	Block, yellow
Z = 4	$0.70 \times 0.34 \times 0.24 \text{ mm}$



Packing diagram for (I), showing $C-H\cdots O$ interactions between molecules A and B. H atoms not involved in the interactions have been omitted.

Data collection

Siemens SMART CCD area- detector diffractometer <i>ω</i> scans Absorption correction: multi-scan (SADAPS: Sheldright 1006)	3147 independent reflections 2797 reflections with $I > 2\sigma(I)$ $R_{int} = 0.015$ $\theta_{max} = 28.3^{\circ}$
$T_{\rm min} = 0.943, T_{\rm max} = 0.980$ 7540 measured reflections	$ \begin{array}{l} n = -11 \rightarrow 11 \\ k = -8 \rightarrow 10 \\ l = -22 \rightarrow 18 \end{array} $
Refinement	
Refinement on F^2 $P[F^2 > 2\sigma(F^2)] = 0.026$	$w = 1/[\sigma^2(F_o^2) + (0.0566P)^2 + 0.1241P]$

$R[F^2 > 2\sigma(F^2)] = 0.036$
$wR(F^2) = 0.100$
S = 1.04
3147 reflections

3147 reflections324 parametersH-atom parameters constrained

Table 1

Selected geometric parameters (Å).

O1A-C12A	1.224 (2)	O1B-C12B	1.219 (3)
O2A - C7A	1.225 (2)	O2B-C7B	1.215 (3)
C1A - C6A	1.424 (3)	C1B-C6B	1.401 (3)
C1A - C12A	1.500 (3)	C1B-C12B	1.501 (3)
C6A - C7A	1.494 (3)	C6B-C7B	1.516 (4)
C7A - C8A	1.463 (3)	C7B-C8B	1.460 (3)
C8A-C11A	1.362 (2)	C8B-C11B	1.343 (3)
C11A-C12A	1.459 (3)	C11B-C12B	1.460 (3)

where $P = (F_o^2 + 2F_c^2)/3$

 $(\Delta/\sigma)_{\rm max} = 0.001$

 $\Delta \rho_{\rm max} = 0.17 \ {\rm e} \ {\rm \AA}^{-3}$

 $\Delta \rho_{\rm min} = -0.17 \text{ e } \text{\AA}^{-3}$

Table 2Geometry of C-H···O interactions (Å, °).

$D - H \cdots A$	$D-{\rm H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$C4A - H4A \cdots O1B^{i}$	0.93	2.58	3.387 (3)	145
$C15B - H15D \cdots O2A$	0.97	2.56	3 216 (3)	125

Symmetry code: (i) 1 + x, 1 + y, z.

All H atoms were positioned geometrically and treated as riding on their parent C atoms, with C—H = 0.93–0.97 Å and $U_{\rm iso}({\rm H})$ = $1.2U_{\rm eq}({\rm C})$. The Friedel reflections were merged before the final refinement because of the absence of significant anomalous scattering effects.

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINT* (Siemens, 1996); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 1997); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used

to prepare material for publication: *SHELXTL, PARST* (Nardelli, 1995) and *PLATON* (Spek, 1990).

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References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1–19.
- Nardelli, M. (1995). J. Appl. Cryst. 28, 659.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). SHELXTL. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.
- Siemens (1996). *SMART* and *SAINT*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.

Spek, A. L. (1990). Acta Cryst. A46, C-34.

- Usman, A., Fun, H.-K., Wang, L., Zhang, Y. & Xu, J.-H. (2002). Acta Cryst. E58, 01463–01465.
- Wang, L., Usman, A., Fun, H.-K. Zhang, Y. & Xu, J.-H. (2003). Acta Cryst. E59, o106–o107.